Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Huub Kooijman,^a* Monique M. van Leijen,^b Otto S. Akkerman,^b Friedrich Bickelhaupt^b and Anthony L. Spek^a

^aBijvoet Centre for Biomolecular Research, Department of Crystal and Structural Chemistry, Utrecht University, Padualaan 8, 3584 CH Utrecht, The Netherlands, and ^bScheikundig Laboratorium, Vrije Universiteit, De Boelelaan 1083, 1081 HV Amsterdam, The Netherlands

Correspondence e-mail: h.kooijman@chem.uu.nl

Key indicators

Single-crystal X-ray study T = 298 KMean $\sigma(C-C) = 0.006 \text{ Å}$ R factor = 0.056 wR factor = 0.131 Data-to-parameter ratio = 18.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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$[\eta^2$ -8-(*tert*-Butylimino)-1-naphthyl]bis(η^5 -cyclopentadienyl)titanium(IV)

The title compound, $[Ti(C_5H_5)_2(C_{15}H_{15}N)]$, displays an iminoacyl fragment in an N-outside conformation, indicating that the C=N vector points away from the σ -bonded C atom. Considering only the geometrical centres of all η -coordinated groups, the Ti^{IV} ion adopts a severely distorted tetrahedral configuration. The crystal packing displays a short C-H··· π contact linking the molecules into an infinite chain running along the *b*-axis direction.

Comment

The structure of the title compound, (I), was determined in the course of our investigations on the synthesis and reactivity of small metallacycles. A similar compound has been reported by Campora *et al.* (1995) and a related compound with Zr by Berg & Petersen (1989).



The structure of (I) is shown in Fig. 1, and selected bond lengths and angles are listed in Table 1. The title compound has two η^5 -coordinated cyclopentadienyl (Cp) rings coordinated to the Ti^{IV} ion. The distances between the perpendicular projection of the Ti^{IV} ion on the Cp rings and their ring centroids (ring slippage) are 0.043 and 0.017 Å for the rings containing C16 and C21, respectively. The title compound further displays an iminoacyl fragment in an N-outside conformation, indicating that the C=N vector points away from the σ -bonded atom C2 (Tatsumi *et al.*, 1985). When the Cp rings and the C=N group are represented by their geometrical centres, as is suggested in Fig. 1, the Ti^{IV} ion displays a severely distorted tetrahedral configuration, with X - Ti - Y angles in the range 88.24 (11)–133.16 (9)° (Table 3). In an alternative description, the Ti^{IV} ion takes part in two adjacent titanacycles, the three-membered ring Ti1/C11/N11 and the five-membered ring Ti1/C2/C1/C10/C11. The maximum deviation from the least-squares plane through this last ring system is 0.028 (3) Å for C2. The Ti1-C2 bond is elongated by 0.2 Å compared with its parent compound 1,8naphthalenediyltitanocene (Tinga et al., 1994). The crystal packing displays a short $C-H \cdot \cdot \pi$ contact (Fig. 2 and Table 2), Received 7 June 2006 Accepted 8 June 2006

metal-organic papers



Figure 1

A view of the title compound, showing 30% probability displacement ellipsoids. H atoms have been omitted for clarity.



Figure 2

Section of the crystal packing illustrating the C4–H4··· π interactions, viewed approximately perpendicular to the (102) planes. H atoms, with the exception of H4, have been omitted.

which links the molecules into an infinite chain running along the *b*-axis direction.

Experimental

A solution of 1,8-naphthalenediyltitanocene (0.1 mmol), prepared according to Tinga *et al.* (1994), in pentane (10 ml) was added to a solution of CN-'Bu (0.2 mmol) in toluene (2 ml) at room temperature. After stirring for 3 h, the solvents were evaporated and the residue was extracted with cyclopentane. Brown crystals were obtained after

a week from a solution in cyclopentane, which was kept at 280 K. Spectroscopic analysis: ¹H NMR (200 MHz, room temperature, C₆D₆, reference $C_6D_5H = 7.17$ p.p.m., p.p.m.): δ 7.85 (*m*, 1 H, H3), 7.83 (*m*, 1 H, H9), 7.68 (m, 1 H, H5), 7.58 (m, 1 H, H4), 7.50 (m, 1 H, H7), 7.37 (m, 1 H, H8), 5.20 (s, 10 H, 2Cp), 1.21 (s, 9 H, ^tBu) (at 240 K a similar proton NMR spectrum was measured in CD₂Cl₂); ¹³C NMR (50.3 MHz, room temperature, C_6D_6 , reference $C_6D_6 = 128.0$ p.p.m., p.p.m.): δ 220.2 (d, ³J = 3.3 Hz, C11), 191.2 (m, C2), 158.1 (m, C10), 140.3 (ddd, ${}^{1}J$ = 154.1 Hz, ${}^{2}J$ = 1.2 Hz, ${}^{3}J$ = 7.5 Hz, C3), 137.3 (d, ${}^{3}J$ = 7.6 Hz, C1 or C6), 135.0 (dd, ${}^{3}J$ = 8.0/6.6 Hz, C6 or C1), 132.0 (d, ${}^{1}J$ = 164.0 Hz, C9), 126.9 (dd, ${}^{1}J$ = 150.9 Hz; ${}^{2}J$ = 2.3 Hz, C4), 124.3 (dd, ${}^{1}J$ = 158.4 Hz, ${}^{2}J$ = 1.9 Hz, C8), 121.6 (*dd*, ${}^{1}J$ = 157.3 Hz, ${}^{3}J$ = 8.9 Hz, C7), 121.1 (*ddd*, ${}^{1}J$ = 157.3 Hz, ${}^{3}J$ = 9.4/8.2 Hz, C5), 105.5 (*dm*, ${}^{1}J$ = 171.0 Hz, 2Cp), 59.2 (d, ^{2}J = 4.2 Hz, C12), 29.3 (gsept, ^{1}J = 125.4 Hz, ^{3}J = 4.7 Hz, ^{*t*}Bu); IR (hexane) $\bar{\nu}$ (cm⁻¹) 1726 (s); HRMS (EI) calculated for C₂₁H₁₆N⁴⁸Ti: 330.0764; found: 330.076; MS (EI) m/z (relative intensities): 387 (not observed, M^+), 330 (77, $[M - M^+]$ ^tBu]⁺), 303 (17, [M - CHN-^tBu]⁺), 178 (38, Cp₂Ti⁺), 128 (100, naphthalene, impurity).

Crystal data

 $\begin{array}{ll} [{\rm Ti}({\rm C}_{\rm 5}{\rm H}_{\rm 5})_2({\rm C}_{15}{\rm H}_{15}{\rm N})] & Z = 4 \\ M_r = 387.33 & D_x = 1.297 \ {\rm Mg \ m^{-3}} \\ {\rm Monoclinic}, \ P_{2_1}/c & {\rm Mo \ } {\it K\alpha \ } {\rm radiation} \\ a = 15.0891 \ (12) \ {\rm \mathring{A}} & \mu = 0.44 \ {\rm mm^{-1}} \\ b = 8.4475 \ (12) \ {\rm \mathring{A}} & T = 298 \ {\rm K} \\ c = 15.6025 \ (18) \ {\rm \mathring{A}} & {\rm Block, \ brown} \\ \beta = 94.200 \ (13)^\circ & 0.5 \times 0.4 \times 0.2 \ {\rm mm} \\ V = 1983.4 \ (4) \ {\rm \mathring{A}}^3 \end{array}$

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: none 5661 measured reflections 4533 independent reflections

Refinement

H-atom parameters constrained
$w = 1/[\sigma^2(F_0^2) + (0.0508P)^2]$
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.28 \text{ e} \text{ Å}^{-3}$
$\Delta \rho_{\rm min} = -0.28 \text{ e} \text{ Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Ti1-N11	2.197 (3)	Ti1-C11	2.043 (3)
Ti1-C2	2.292 (3)	N11-C11	1.238 (4)
N11-Ti1-C2	104.45 (11)	C11-N11-C12	132.5 (3)
N11-Ti1-C11	33.70 (12)	N11-C11-C10	150.8 (3)
C2-Ti1-C11	70.75 (12)		

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
$C4-H4\cdots Cg4^{i}$	0.93	2.98	3.768 (5)	143

Symmetry code: (i) -x + 1, $y - \frac{1}{2}$, $-z + \frac{1}{2}$. Cg4 is the centroid of the C1/C6/C7/C8/C9/C10 ring.

2344 reflections with $I > 2\sigma(I)$

3 standard reflections

frequency: 60 min

intensity decay: 1.5%

 $\begin{aligned} R_{\rm int} &= 0.200\\ \theta_{\rm max} &= 27.5^\circ \end{aligned}$

Table 3

Tetrahedral geometry of Ti1 (angles in °).

Cg1 represents the centroid of the Cp ring C16–C20; Cg2 represents the centroid of the Cp ring C21–C25; Cg3 represents the mid-point of the N=C bond.

C2-Ti1-Cg1	103.22 (10)
C2-Ti1-Cg2	101.46 (11)
C2-Ti1-Cg3	88.24 (11)
Cg1-Ti1-Cg2	133.16 (9)
Cg1-Ti1-Cg3	110.35 (9)
Cg2-Ti1-Cg3	109.60 (10)

H atoms were introduced in calculated positions, with C—H = 0.93 and 0.96 Å, and refined as riding on their carrier atoms, with $U_{\rm iso}({\rm H}) = 1.5 U_{\rm eq}({\rm methyl C})$ or $1.2 U_{\rm eq}({\rm other C})$. The methyl groups were refined as rigid groups, allowing for rotation around the C—C bond.

Data collection: locally modified *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *SET4* (de Boer & Duisenberg, 1984); data reduction: *HELENA* (Spek, 1997); program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985); program(s) used to refine struc-

ture: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

This work was supported in part (ALS) by the Council for the Chemical Sciences of the Netherlands Organization for Scientific Research (CW-NWO).

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